

Resonance Topology of Polynuclear Aromatic Hydrocarbons

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Equations for the determination of the number of unexcited resonance forms (N) for several lattice systems of polynuclear aromatic hydrocarbons have been developed, e.g., for the square (parallelogram) model, $S[m, n]$, the symmetric circular systems $C[m, n]$, and the rectangular model $R[m, n]$, using combinatorial mathematics. Equations for the three major classes (R, C, S) of polynuclear aromatic hydrocarbons can be reduced to a generalized formula.

An additional form, the skew strip model, $Z[m, n]$, is also described, for which N may be determined by an independent formula. However, an addition of one ring to the $R[m, n]$ series will result in a $Z[m+1, n]$ series. Thus, ring addition may transfer a lattice from one geometrical system to another, both subject to description by the same set of internally consistent equations for the calculation of N . The successive coupling of benzene rings to a specific series is general, so that an explicit formula for N of an intermediate lattice, e.g., the peropyrene type molecule, $PP[m, n]$, can be derived from a known model type, e.g., $C[m, n]$. The four major models (R, C, S , and Z) discussed are descriptive of most polynuclear aromatic hydrocarbons frequently encountered.

Gordon and Davison [1] have calculated the number of unexcited forms, N , of the square (parallelogram) model, $S[m, n]$ (Fig. 1a),

$$N = m+n C_m \quad (1)$$

and the symmetric circular model (hexagonal assemblages of hexagons) $C[m, n]$, $m = n$ (Fig. 1b)

$$N = \frac{\prod_{a=1}^{2n-1} n+a C_n}{\prod_{a=1}^{n-1} n+a C_n} \quad (2)$$

from the consideration of the possible paths in a dot diagram of the hexagonal lattice.

Based on combinatorial mathematics, we have developed a formula for the N of the rectangular model [2], $R[m, n]$ polynuclear aromatic hydrocarbons (Fig. 1c)

$$N = (n+1)^m \quad (3)$$

e.g., for the polyphenylenes, $R[m, 1]$, N is 2^m ; and for the polyacenes, $R[1, n]$, N is $n+1$. The polyrylene series is $R[m, 2]$ and N is 3^m . This series includes hydrocarbons such as perylene ($m=2$), terylene ($m=3$), and quarterylene ($m=4$) etc. The polyanthene series is $R[m, 3]$ and N is 4^m , which includes hydrocarbons

such as bisanthene ($m=2$) tetraanthene ($m=3$), etc. For hydrocarbons of the circular model, $C[m, n]$.

$$N = \frac{\binom{m+n}{n} \binom{m+n+1}{n} \dots \binom{m+n+(m-1)}{n}}{\binom{n+1}{n} \dots \binom{n+(m-1)}{n}} \quad (4)$$

This expression is also applicable for circular hydrocarbons, $m \neq n$, whose shape deviates from that of an array of the hexagonal assemblages. For the square model, $S[m, n]$ or $S[n, m]$, Eq.(1) can be written as

$$N = \binom{m+n}{n}. \quad (5)$$

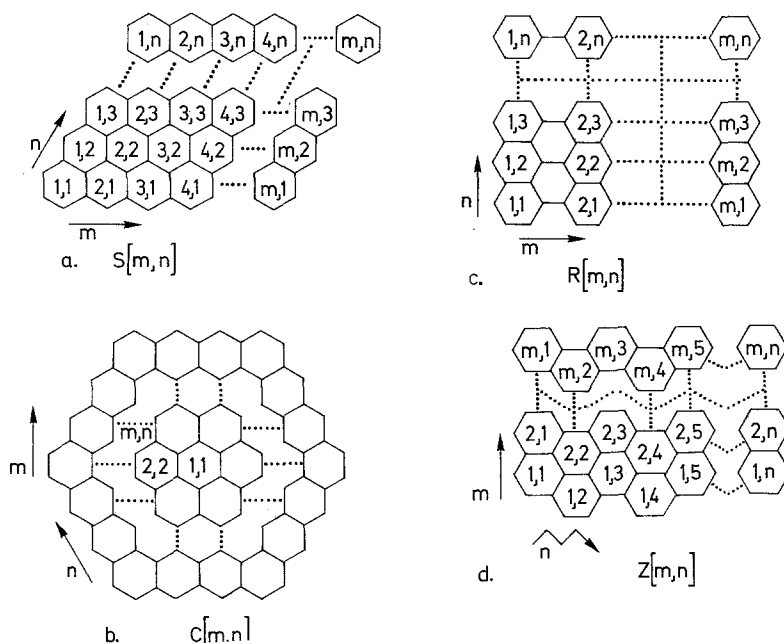


Fig. 1. a The square or parallelogram model, $S[m, n]$, where m and n are interchangeable. $S[1, 1]$ is benzene, $S[2, 1]$ or $S[1, 2]$ is naphthalene, $S[2, 2]$ is pyrene, etc. Both m and n units expressed are in the naphthalene type linkage. Hydrocarbons of $C = 2[m(n+1) + n]$, $H = 2(m+n+1)$ belong to this class. b The circular model $C[m, n]$ where m is expressed in units of the biphenyl type linkage along the axis (y) perpendicular to the horizontal acene type axis (x) and n is along the 4-edges of the molecule. $C[1, 1]$ is benzene, $C[2, 2]$ is coronene, $C[2, 1]$ is pyrene, $C[1, 2]$ is naphthalene, $C[2, 3]$ is ovalene, etc. Hydrocarbons of $C = 2m(m+2n)$ and $H = 2(2m+n)$ belong to this class. c The rectangular model $R[m, n]$ originated by Coulson (Ref. 2). Index m indicates units of the biphenyl type linkage along the x -axis and n , the naphthalene type linkage along the y -axis. $R[1, 1]$ is benzene, $R[2, 1]$ is naphthalene, $R[1, 2]$ is biphenyl, $R[2, 2]$ is perylene, etc. Hydrocarbons of $C = 2m(2n+1)$ and $H = 2(2m+n)$ belong to this model. d The skew strip model is $Z[m, n]$. Index m is units of naphthalene type linkage along the zig-zag path of the x -axis and n is that along the straight-line path of the y -axis. $Z[1, 1]$ is benzene, $Z[2, 1]$ is naphthalene, $Z[3, 1]$ is phenanthrene, $Z[2, 2]$ is pyrene, $Z[3, 2]$ is 1,12-benzoperylene, etc. See Ref. [3]

The above Eqs. (3), (4), and (5) can be reduced to a generalized formula for three major classes (*R*, *S*, and *C*) of polynuclear aromatic hydrocarbons.

$$N = \binom{n+b}{n}^c \prod_{q=1}^{m-1} \binom{m+n+q}{n} \Big/ \binom{n+a+q}{n}$$

for

$$R[m, n]: a = m, b = 1, c = m,$$

$$S[m, n]: a = m, b = m, c = 1, \quad (6)$$

$$C[m, n]: a = 0, b = m, c = 1.$$

Another important model is the skew strip, $Z[m, n]$ series (Fig. 1d). For $Z[m, 1]$, N takes the form of the Fibonacci series, 2, 3, 5, 8, 13 ... or the recurrence form, $f(0) = 0, f(1) = 1, f(k) = f(k-1) + f(k-2)$ for $k = 2, 3 \dots$ and $m+2 = k$. This series also can be expressed by the golden section ratio $\tau, 1 + \tau^{-1} = \tau$ or $\tau^2 = 1 + \tau$, upon solving $\tau = (a+1)/2$ where $a = 5^{1/2}$; i.e., $\tau^k = f(k-1) + f(k)\tau$ and $(-\tau)^{-k} = f(k-1) + f(k)(1-\tau)$. Therefore, $f(k) = \frac{\tau^k - (-\tau)^{-k}}{\tau + \tau^{-1}}$ which can be written as

$$N = a^{-1} 2^{-(m+2)} [(1+a)^{m+2} - (1-a)^{m+2}]. \quad (7)$$

This form can be modified by expansion of $(1+a)^{m+2}$ and $(1-a)^{m+2}$ based on Newton's formula and then simplified as

$$N = 2^{-(m+1)} \prod_{k=1}^{(m+2)/2} \binom{m+2}{2k-1} 5^{(k-1)}, \quad m = \text{even},$$

$$N = 2^{-(m+1)} \sum_{k=1}^{(m+3)/2} \binom{m+2}{2k-1} 5^{(k-1)}, \quad m = \text{odd}. \quad (8)$$

The above expression is a general form for N with any m for $Z[m, 1]$. On the other hand, the $Z[1, n]$ series is identical to the polyacene series $R[1, n]$. The two-tier type, the $Z[2, n]$, is the series containing naphthalene, $Z[2, 1]$, pyrene, $Z[2, 2]$, anthanthrene $Z[2, 3]$, etc. For this series, N can be calculated from $\sum_{k=1}^{n+1} k$, and, after expansion, N is given by $(n+1)(n+2)/2$ which is essentially $\binom{n+2}{n}$, derived from Eq. (6) from $S[2, n]$.

The next sequence, $Z[3, n]$, has a three tier strip with one end indented and is pentagonally shaped (in the form of chevron). $Z[3, 1]$ is phenanthrene; $Z[3, 2]$ is 1,12-benzoperylene, etc. Derivation of N yields

$$N = \sum_{k=1}^{n+1} k^2 = (n+1)(n+2)(2n+3)/6. \quad (9)$$

Eq. (9) can also be derived by adding one benzene ring to the $R[2, n]$ series, e.g., the addition of one benzene ring to biphenyl, $R[2, 1]$, yields phenanthrene, $Z[3, 1]$; adding, in the same manner, one benzene ring to perylene, $R[2, 2]$, becomes 1,12-benzoperylene, $Z[3, 2]$; etc.

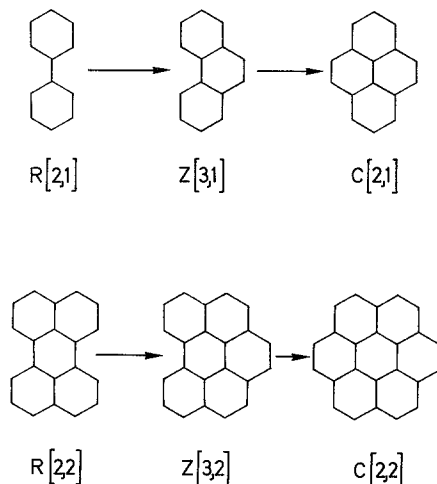


Fig. 2

In general, the addition of one benzene ring to the $R[2, n]$ series will result in the $Z[3, n]$ series. The formula for the $R[2, n]$ series is $N = (n + 1)^2$. When one ring is coupled to $R[2, n]$, Eq.(10) is obtained for $Z[3, n]$.

$$\begin{aligned}
 N &= 1 + \sum_1^n (n + 1)^2 \\
 &= 1 + n(n + 1)(2n + 1)/6 + n(n + 1) + n \\
 &= (n + 1)(n + 2)(2n + 3)/6.
 \end{aligned} \tag{10}$$

To verify Eq. (10), another benzene ring is coupled to the $Z[3, n]$ series to produce the $C[2, n]$ series, e.g., from phenanthrene, $Z[3, 1]$ to pyrene, $C[2, 1]$; from 1,12-benzperylene, $Z[3, 2]$, to coronene, $C[2, 2]$

$$\begin{aligned}
 N &= 1 + \sum_1^n (n + 1)(n + 2)(2n + 3)/3! \\
 &= (n + 1)(n + 2)^2(n + 3)/12 \\
 &= \binom{n + 2}{n} \binom{n + 3}{n} / \binom{n + 1}{n}.
 \end{aligned} \tag{11}$$

Eq. (11) gives a formula for $C[2, n]$ identical to that obtained from Eq. (6).

Apparently the successive coupling of the benzene ring to a specific series is quite general, e.g., if we want to write an explicit form of N for the 5-tier strip, i.e., the peropyrene type of molecule, $PP[m, n]$, we could derive it from a known model type, e.g., the C -model.

From Eq. (6), the N for $C[3, n]$ should be
$$\frac{\binom{n + 3}{n} \binom{n + 4}{n} \binom{n + 5}{n}}{\binom{n + 1}{n} \binom{n + 2}{n}}.$$

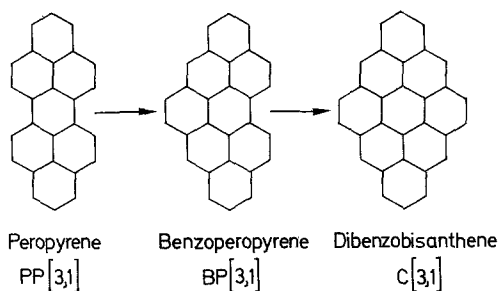


Fig. 3

After considering the numbers of $n=1$ to $n=3$, one can construct a table predicting the values for the $PP[3, n]$ series.

n	$PP[3, n]$	$BP[3, n]$	$C[3, n]$
1	18	19	20
2	136	155	175
3	650	805	980

By inspection, one can deduce the column of values from right to left as $1 + 19 = 20$; $1 + 19 + 155 = 175$; $1 + 19 + 155 + 805 = 980$; etc. Hence we can write $BP[3, n]$ as

$$\frac{\binom{n+3}{n} \binom{n+4}{n} \binom{n+5}{n}}{\binom{n+1}{n} \binom{n+2}{n}} - \frac{\binom{n+2}{n-1} \binom{n+3}{n-1} \binom{n+4}{n-1}}{\binom{n}{n-1} \binom{n+1}{n-1}}$$

or for the $PP[3, n]$ series,

$$\frac{\binom{n+3}{n} \binom{n+4}{n} \binom{n+5}{n}}{\binom{n+1}{n} \binom{n+2}{n}} - \frac{2 \binom{n+2}{n-1} \binom{n+3}{n-1} \binom{n+4}{n-1}}{\binom{n}{n-1} \binom{n+1}{n-1}} + \frac{\binom{n+1}{n-2} \binom{n+2}{n-2} \binom{n+3}{n-2}}{\binom{n-1}{n-2} \binom{n}{n-2}}$$

which can be abbreviated as

$$\begin{aligned}
 N_{PP[3, n]} &= \sum_{m=1}^3 \binom{n+m}{n+m-3} \prod_{q=1}^2 \binom{n+m+q}{n+m-3} \Big/ \binom{n+m+q-3}{n+m-3} \\
 &\quad - 3 \binom{n+2}{n-1} \sum_{q=1}^2 \binom{n+q+2}{n-1} \Big/ \binom{n+q-1}{n-1} \\
 &= (n+1)(n+2)^3(n+3)(n^2+4n+5)/5! .
 \end{aligned}
 \tag{12}$$

It can be shown that

$$N_{BP[3, n]} = 1 + N_{PP[3, n]} . \tag{13}$$

and

$$N_{C[3, n]} = 1 + N_{BP[3, n]} . \tag{14}$$

The four major models (*R*, *C*, *S*, and *Z*) discussed will cover most polynuclear hydrocarbons frequently encountered, either kata- or pericondensed. Physically, since *N* is related to certain resonance forms with a specified degree of excitation, it is to be anticipated that *N* will correlate with a number of properties such as delocalization energy, boiling point, etc.

N-values obtained for a few aromatic hydrocarbons of the four models are included in the Table.

Table. *N*-Values of some polynuclear aromatic hydrocarbons

Shape	Hydrocarbon	<i>N</i>
<i>R</i> [1, 11]	Undecacene	12
<i>C</i> [2, 2]	Coronene	20
<i>R</i> [2, 2]	Perylene	9
<i>C</i> [2, 4]	Circumanthracene	105
<i>R</i> [6, 1]	Sexiphenyl	64
<i>S</i> [3, 3]	5,6,12,13-Dibenzoperopyrene	20
<i>Z</i> [5, 1]	Picene	13
<i>R</i> [4, 2]	Quaterylene	81
<i>C</i> [3, 3]	Circumcoronene	980
<i>Z</i> [3, 2]	1,12-Benzoperylene	14
<i>C</i> [2, 3]	Ovalene	50
<i>C</i> [4, 4]	—	232, 848

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